

What is a **CATALYST**?



A **catalyst** is a substance that accelerates or facilitates a chemical reaction without being consumed in the process.

It acts as a "push" to make the reaction occur more quickly or under more favorable conditions.

The **catalyst** provides an alternative pathway for molecules to bond or break apart, lowering the activation energy required.

It serves as a facilitator that helps the chemical reaction proceed more efficiently.

Once the reaction is complete, the catalyst still exists and can be used again in future reactions.

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J Rink



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