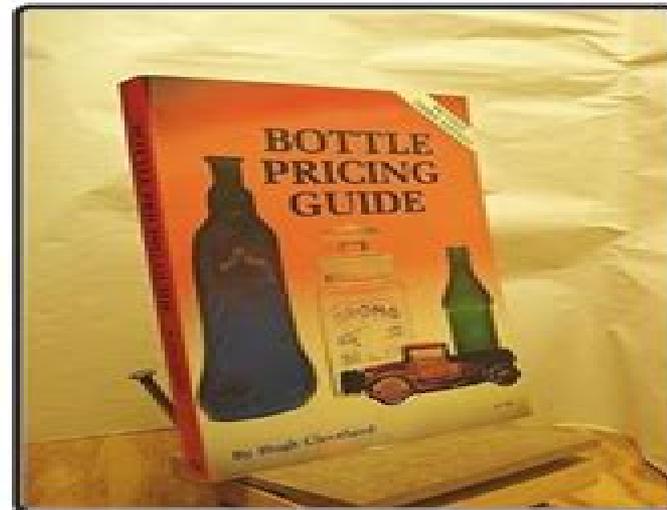


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 $\text{H}_2\text{O} + \text{KBr} \rightarrow \text{CO}_2 + \text{H}_2\text{O} + \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + 6\text{NaBr}$
br using the following equation
 $2\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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 $\text{N}_2 + 3\text{F}_2 \rightarrow 2\text{NF}_3$
 $\text{C}_6\text{H}_{10} + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$
 $\text{H}_2\text{O} + \text{KBr} \rightarrow \text{CO}_2 + \text{H}_2\text{O} + \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + 6\text{NaBr}$
 $\text{H}_2\text{O} + \text{KBr} \rightarrow \text{CO}_2 + \text{H}_2\text{O} + \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + 6\text{NaBr}$
 $\text{H}_2\text{O} + \text{KBr} \rightarrow \text{CO}_2 + \text{H}_2\text{O} + \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + 6\text{NaBr}$

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answer $3\text{C} + \text{SiO}_2 \rightarrow \text{SiC} + 2\text{CO}$

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 $\text{N}_2 + 3\text{F}_2 \rightarrow 2\text{NF}_3$
 $\text{C}_6\text{H}_{10} + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$
 $\text{H}_2\text{O} + \text{KBr} \rightarrow \text{CO}_2 + \text{H}_2\text{O} + \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + 6\text{NaBr}$
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same thing goes with the o3 i am so lost

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