

Review of Basic Mathematics

Whole Numbers

Addition and Subtraction

Addition is indicated by $+$. Subtraction is indicated by $-$. Commutative is a special mathematical name we give to certain operations. It means that we can do the operation in any order. Addition is commutative because we know that

$$2 + 4 \text{ means the same thing as } 4 + 2$$

Subtraction is not commutative since

$$21 - 6 \text{ is not the same as } 6 - 21$$

For this reason addition can be done in any order. Generally we calculate from left to right across the page. We can sometimes rearrange the order of a sum involving both additions and subtractions but when we do this we must remember to keep the number with the sign immediately preceding it. For example

$$3 + 5 + 3 + 4 + 7 + 3 = 3 + 3 + 3 + 4 + 5 + 7$$

$$6 + 7 - 10 + 2 - 1 - 2 = 6 + 7 + 2 - 10 - 1 - 2$$

Multiplication and Division

Multiplication is indicated by \times or \cdot . Sometimes, as long as there will be no confusion we use juxtaposition to indicate multiplication, particular when we use letters to represent quantities. For example $3a$ means $3 \times a$ and xy means $x \times y$. Multiplication is a commutative operation i.e. we can reverse the order e.g. $3 \times 4 = 4 \times 3$.

Division is indicated by the symbols \div , $/$ or --- . Division is not commutative since $4 \div 2 \neq 2 \div 4$. Multiplication and division should be done from left to right across the page, although you can rearrange the order of an expression involving only multiplications. For example

$$3 \times 4 \times 5 = 5 \times 4 \times 3$$

$$ab3a = 3aab$$

Basic Mathematics A Review

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vocabulary to explain scientific concepts. 2.I CAN Distinguish between fission ... Matter and Change • Chapter 25 When a radioactive nucleus gives off a gamma ray, its atomic number increases by 1. The three types of radiation were first identified by Ernest Rutherford. Nuclear Chemistry - Lake Central High School Jul 12, 2015 — What is the change in atomic number after the alpha decay? It decreases by 2. b. ... answer the following questions. **Nuclear** ... 25.2 Nuclear Transformations | Lecture notes Chemistry These nuclei decay by turning a neutron into a proton to emit a beta particle (an electron) from the nucleus. This process is known as beta emission. It ... 60 s - 1 min SECTION 25.2 NUCLEAR TRANSFORMATIONS. 1. Write a nuclear equation for the following radioactive processes. a. alpha decay of francium-208 ${}^{208}\text{Fr} \rightarrow \text{b} \dots$