

for Biology: Concepts & Connections, Sixth Edition

by Campbell, Reece, Taylor, Simon, and Dickinson

Chapter 1: Biology: Exploring Life

- Levels of biological organization from smallest to largest: molecules → organelles → cells → tissues → organs → organ systems → organisms → population → community → ecosystem → biosphere
- Some organisms consist of a single level of organization. The complexity of life increases at the cellular level. In all levels, structure correlates with function.
- The flow of life's energy and diversity in the genetic information is DNA's heredity.
- Biology explores life with three dimensions: (1) structure and (2) function are inseparable phenomena; organ systems (3) behavior includes genetic, physical, and mental. Structure and function are closely intertwined. The physical form and function are closely intertwined. The chemical structure of molecules, tissues, and organs is related to the form and function of the organism.
- Life performs four essential functions: form, energy, and information.



Chapter 2: The Chemical Basis of Life

- Matter consists of elements. Elements are combined to form molecules.
- Twenty-two elements are essential for life. Oxygen, carbon, hydrogen, and nitrogen are the most abundant.
- Atoms are the smallest units of matter that retain the properties of an element. Atoms consist of protons (+), electrons (-), and neutrons (0). Changes in electron number of an atom result in forming ions of an element.
- Electron arrangement determines the chemical properties of an atom. Atoms with incomplete outer shells gain, lose, or share electrons, forming chemical bonds with other atoms.
- Some bonds are attractive forces that oppositely charged ions.
- Atoms are neutral, bonds are polar. Some water exhibits skills by sharing electrons.
- Water is a polar molecule. Unequal electron sharing results in a slightly negative (δ-) end and two slightly positive (δ+) ends.



- Weak attractions between slightly charged (δ-) and oppositely charged (δ+) ends of molecules are called hydrogen bonds. Water's polarity and hydrogen

bonding explain most of its life-supporting properties.

- Water's cohesion has many (δ-) ends than (δ+) ends and a pH less than 7. Water's adhesion has strong (δ-) ends than (δ+) ends and a pH greater than 7.
- Chemical reactions change the composition of matter as reactants are converted to products.

Chapter 3: The Molecular Basis of Cells

- Carbon is at the center of life's chemistry. Organic compounds vary in (1) their size and shape of the carbon skeleton and (2) the presence and location of functional groups.
- Cells make a large number of polymers from a small set of monomers, linked by dehydration reactions.
- Polymers have a linear backbone, generally with each energy and molecular building blocks. Polymers have many ways to animals, including (1) plants can make and provide structure to plant cell walls, (2) cells.
- Lipids are hydrophobic. They include fat energy storage, phospholipids cell membranes, waxes, and steroids. Lipids are hydrophobic and non-polar. Most animal fats are saturated. Plant oils are usually unsaturated. Some long hydrocarbon chains of lipids benefit in the surface display.
- Proteins are important structural elements of cells and participate in most cellular activities. Proteins are linear amino acids are the monomers of proteins. The carboxyl end of one amino acid is linked to a specific bond with the amino group of the next to form a polypeptide.
- A protein consists of one or more polypeptide chains. Each chain is a specific shape that determines its protein function. Four levels may contribute to this shape: (1) the amino acid sequence (primary structure), (2) folding or globular form (secondary structure), (3) the overall shape (tertiary structure), and (4) association of more than one polypeptide (quaternary structure).
- Nucleic acids are polymers of nucleotides. DNA, the molecule of inheritance, is a double helix. Two polynucleotide strands (sugar-phosphate backbone) are linked together by hydrogen bonds between bases. DNA is transcribed into RNA, a single-stranded molecule, which is then translated into the primary structure of proteins.

Chapter 4: A Tour of the Cell

- Prokaryotic cells are small and structurally simple.



- Eukaryotic cells are larger and more complex, with a nucleus and membrane-bounded organelles that perform specific functions.

Chapter 5: The Animal Cell



Chapter 6: The Plant Cell



- The plasma membrane and the extensive network endoplasmic reticulum forms a continuous layer placed on the membrane structure. Phospholipids are the membrane's main components.
- Water, the most abundant (70%) and protein make up elements. Each flow is a characteristic.
- The endoplasmic reticulum (ER) consists of a network of membranes through ER makes phospholipids and various proteins. Smooth ER synthesizes lipids and acts as detoxification.
- Chloroplasts are the rough ER and nuclear envelope, and in the cytoplasm, make proteins.
- The Golgi apparatus contains ER products and packages them to transport within.
- Lysosomes, two membrane digestive systems function as digestive within the cell and defend against it.
- Plant cells have cell walls, a large central vacuole, chloroplasts, which convert solar energy to chemical energy, and photosynthesis's function between adjacent cells.
- Mitochondria, double-membraned organelles, are the sites of cellular respiration, converting food energy to ATP. Mitochondria, the organelles, evolved by endosymbiosis.
- Multicellular organisms, tissues, and some animals are present. Some cells make up the extracellular matrix, cells communicate with each other.
- Cells and tissues are made up of specialized cells, groups of cells.
- Protein between animal cells help hold them together, and some provide communication between cells.

Chapter 7: The Working Cell

- A cell is a unit of genetic function. It is a specialized cell. The protein work is structural support, nutrient activity, cell wall synthesis, signal transduction, formation of cell junctions, and transport.

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