

Basic Nuclear Physics

- An atom consists of a small, positively charged nucleus surrounded by a negatively charged cloud of electrons
- Nucleus
 - Positive protons
 - Neutral neutrons
 - Bond together by the strong nuclear force
 - Stronger than the electrostatic force binding electrons to the nucleus or repelling protons from one another
 - Limited in range to a few $\times 10^{-15}$ m
- Because neutrons are electrically neutral, they are unaffected by Coloumbic or nuclear forces until they reach within 10^{-15} m of an atomic nucleus
 - Best particles to use for FISSION

Basic Nuclear Physics

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